



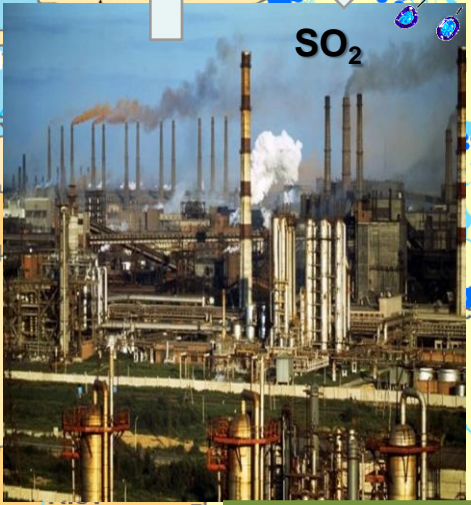
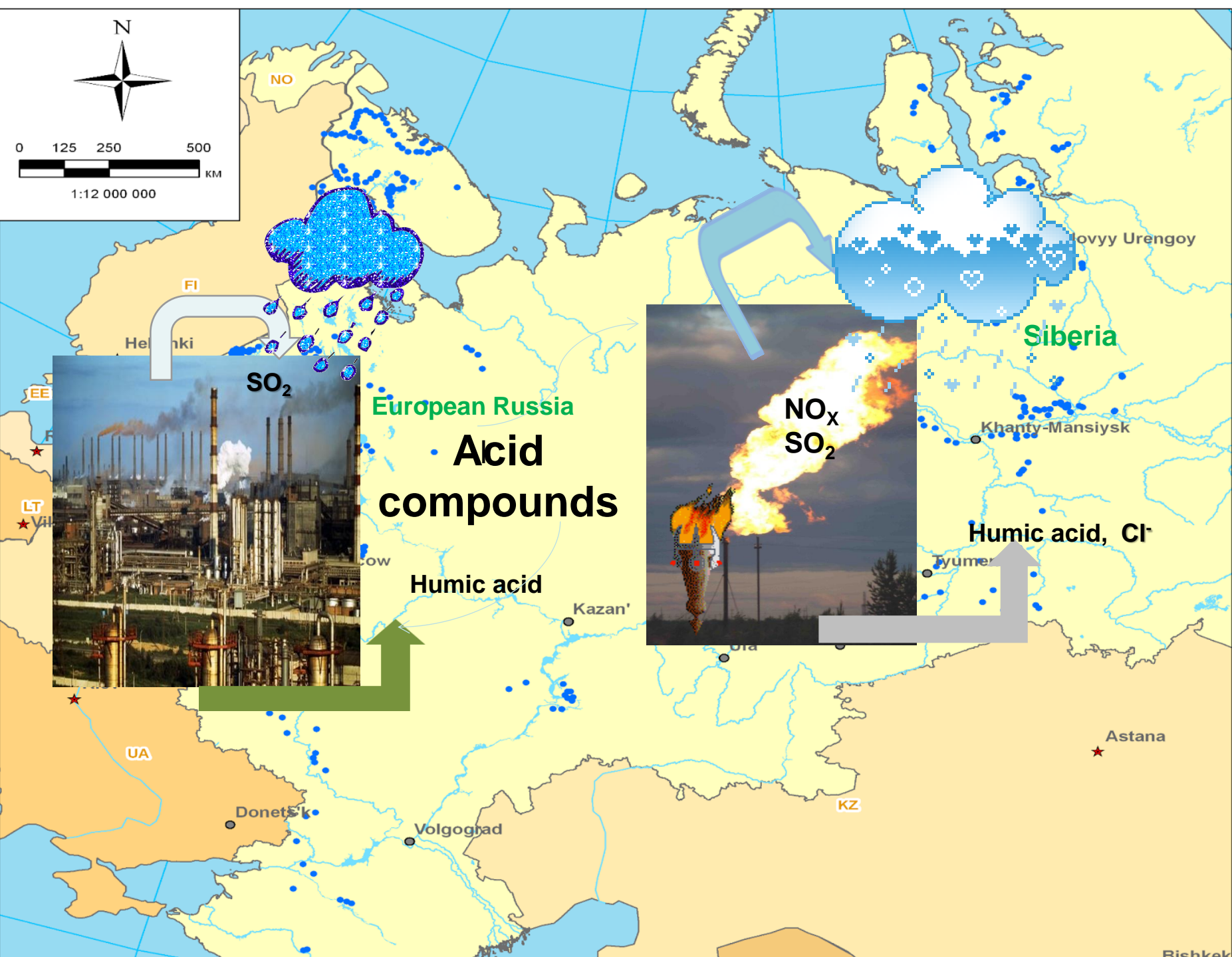
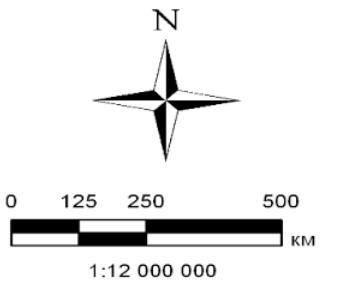
V.I.Vernadsky Institute of
Geochemistry and Analytical
Chemistry RAS



State University of Tyumen

Pollution impact on lakes quality of Russian Arctic region

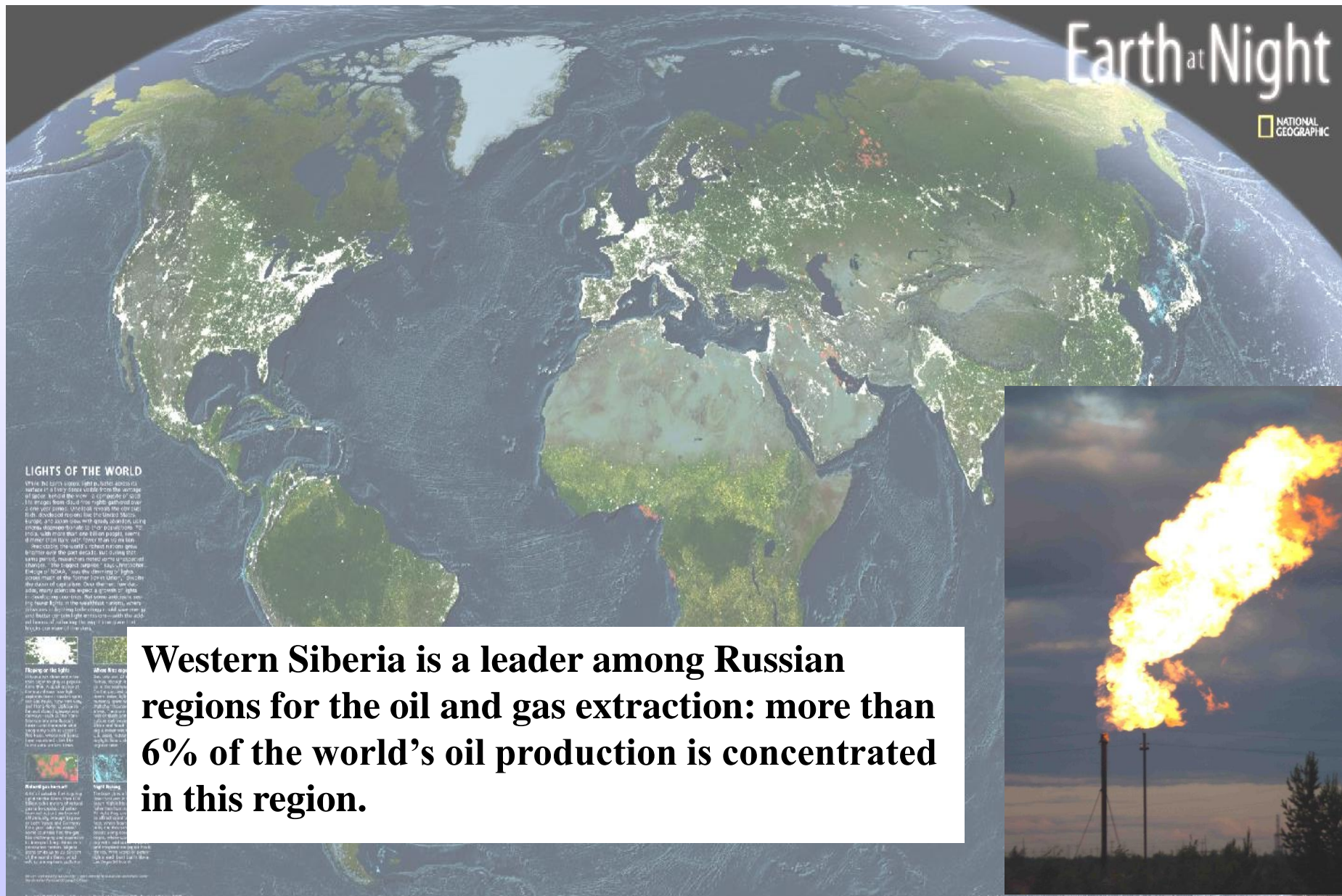
***Moiseenko T.I., Dinu M.I.,
Kremleva T.A., Gashkina N.A.***



European Russia
Acid compounds
Humic acid

Siberia
Humic acid, Cl⁻

Gas flaring during of oil production in Western Siberia leads to air pollution by oxides of nitrogen, sulfur, chlorine.



Earth at Night

NATIONAL GEOGRAPHIC

LIGHTS OF THE WORLD

While the Earth's oceans, hills, and valleys are mostly dark at night, the world's cities and towns are lit up. The lights of the world are a mix of natural and artificial light. Natural light comes from the sun, moon, and stars. Artificial light comes from cities, towns, and villages. The lights of the world are a mix of natural and artificial light. Natural light comes from the sun, moon, and stars. Artificial light comes from cities, towns, and villages.

Flaring of the lights
The lights of the world are a mix of natural and artificial light. Natural light comes from the sun, moon, and stars. Artificial light comes from cities, towns, and villages.

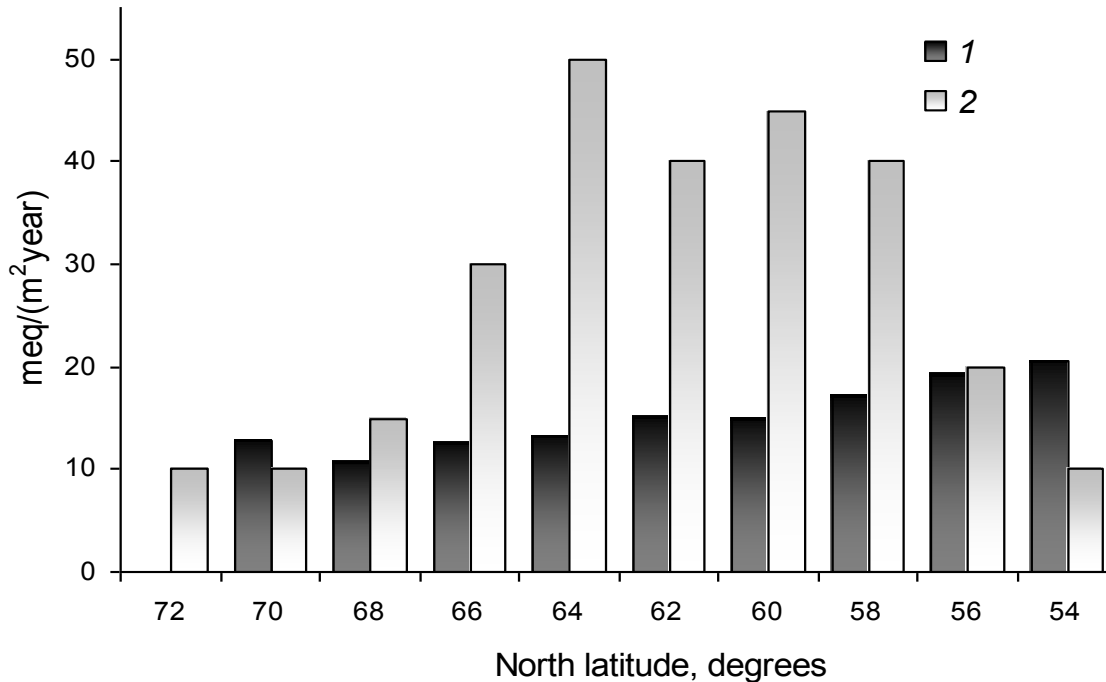
World gas flaring
The lights of the world are a mix of natural and artificial light. Natural light comes from the sun, moon, and stars. Artificial light comes from cities, towns, and villages.

Night flaring
The lights of the world are a mix of natural and artificial light. Natural light comes from the sun, moon, and stars. Artificial light comes from cities, towns, and villages.

Western Siberia is a leader among Russian regions for the oil and gas extraction: more than 6% of the world's oil production is concentrated in this region.



Deposition



Deposition of strong acids (meq/m²·year) by the latitudinal gradient **in European territory of Russia - 1** (EMEP, 2000) and **in Western Siberia - 2** (Semenov, 2002)

pH of precipitation

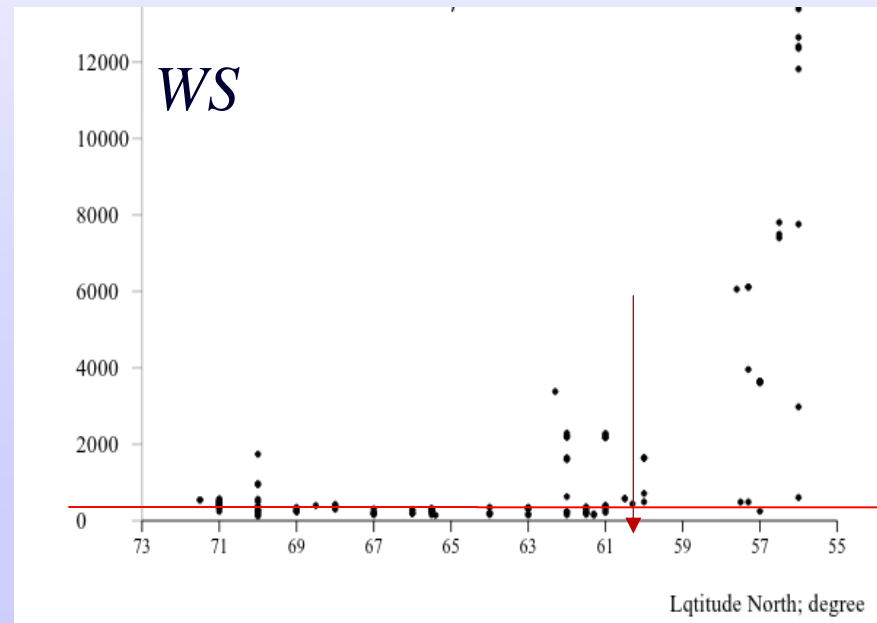
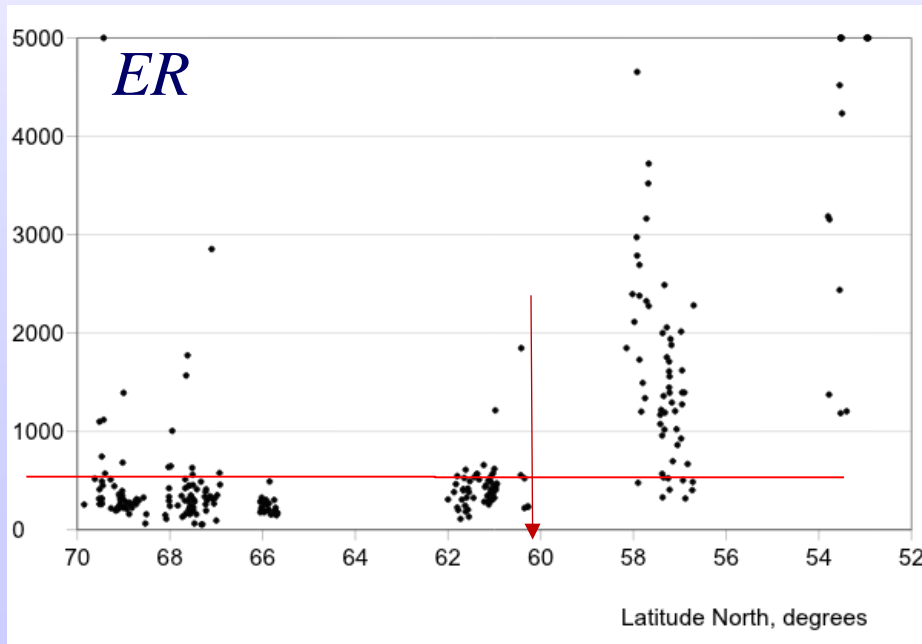
Region	pH of precipitation	
	Min	Max
North and North-West of ER	3.1	6.2
Center of ER	3.2	7.0
South of ER	3.1	7.1
Urals and the Urals region	4.0	7.2
Center of Western Siberia northern coast and north-eastern seas	3.6	7.0

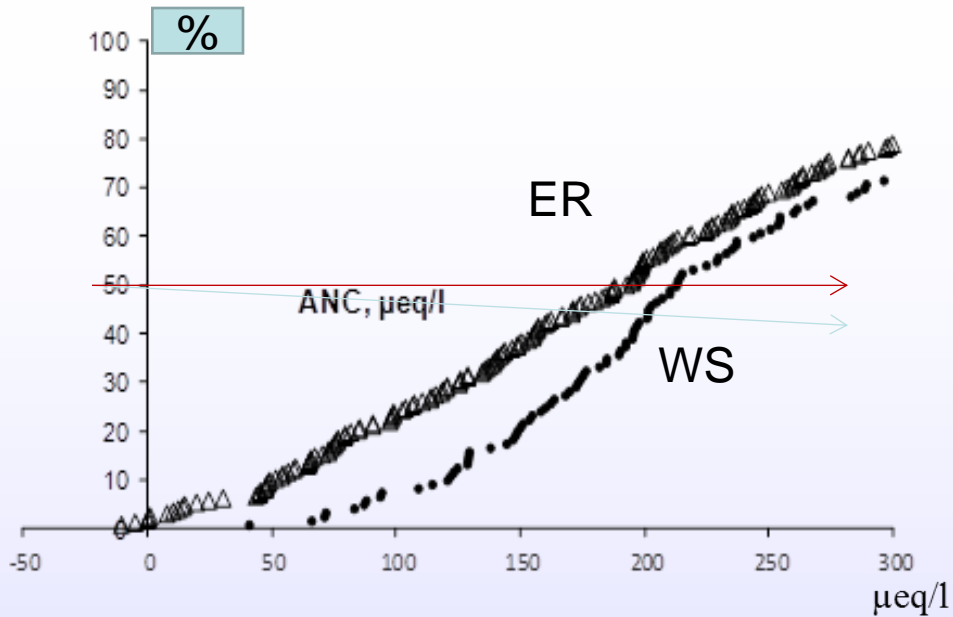
Buffer capacity

The main feature of both regions is **an increase of cations and alkalinity concentrations** in water towards the south: for ER - 55-60 °, for WS - 55-60 ° North latitude. The lakes of forest-steppe zones in ER and WS are highly resistant to acidification.

Buffer capacity of northern and middle taiga region of ER and WS connect with the features of geology.

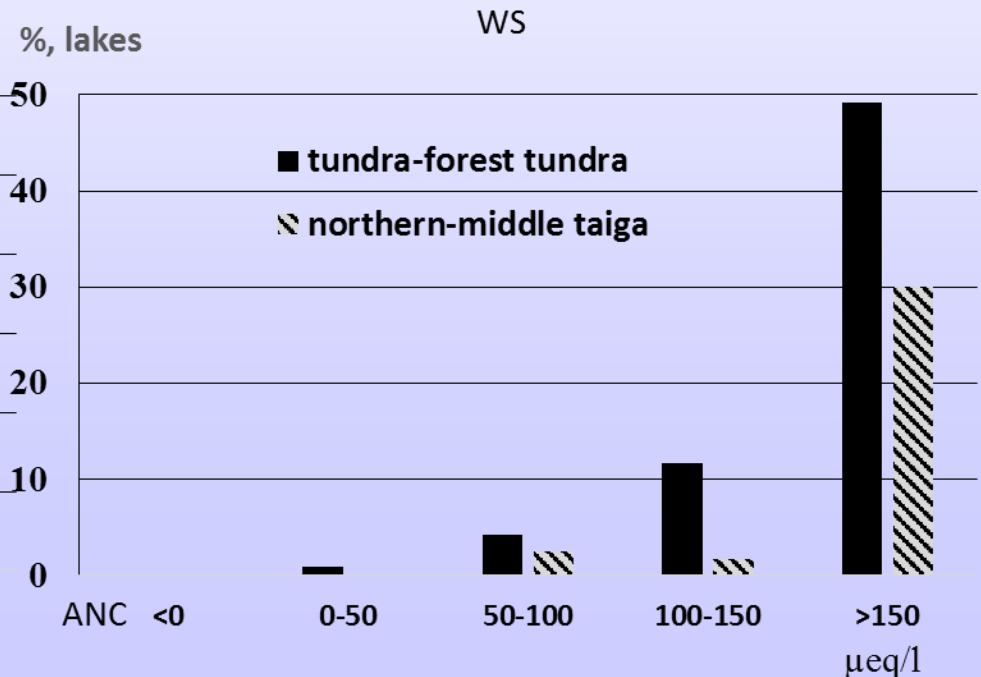
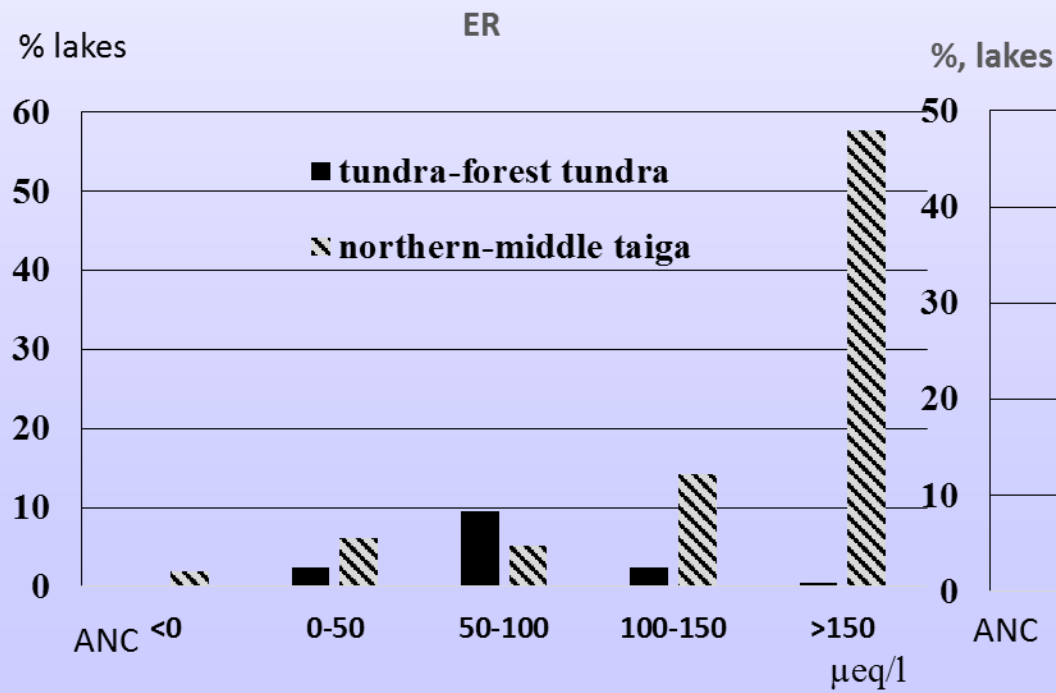
Concentration of cations (Ca+Mg+K+Na), $\mu\text{eq/l}$, in smalls lakes



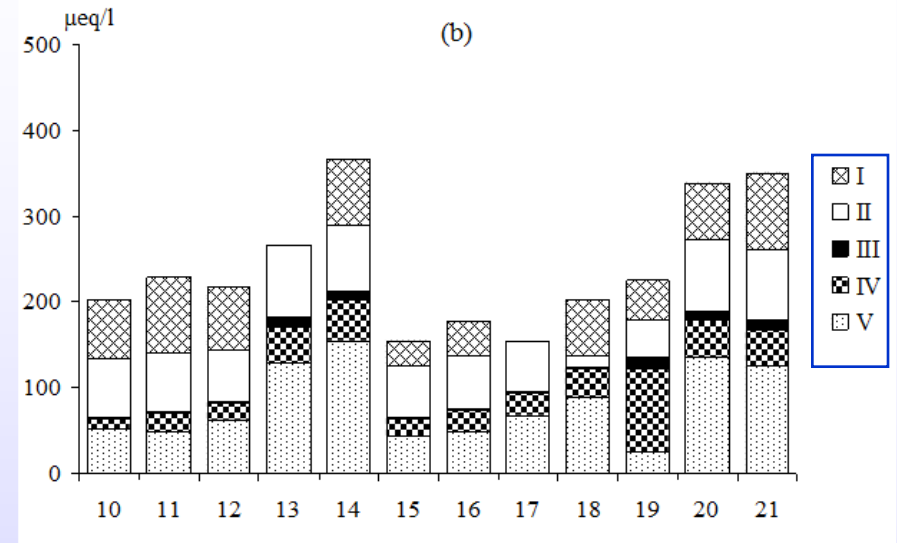
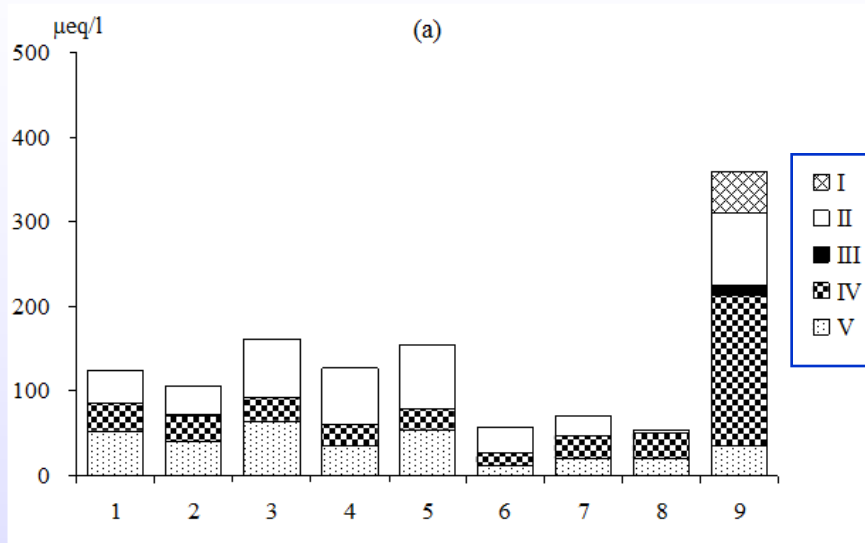


The acid neutralizing capacity of water (ANC, $\mu\text{eq/l}$) is used to estimate the anthropogenic acidification as a difference between cations and anions of strong acids (Henriksen et. al., 1992)

1. $\text{ANC1} = \text{Ca}^{2+} + \text{Mg}^{2+} + \text{Na}^{+} + \text{K}^{+} - \text{SO}_4^{2-} - \text{NO}_3^{-}$
2. $\text{ANC2} = \text{HCO}_3^{-} + \text{A}^{\text{n}} - \text{H}^{+} - \text{Al}^{3+}$



The anionic composition (I – Alk. II - A^n . III – NO_3^- . IV – SO_4^{2-} . V – Cl^-) of the water lakes with **pH < 5** on the ER (a) (1 - in the tundra, 2-8 - in the northern taiga, 9 - in the middle taiga) and WS (b) (10-12 in the tundra, 13-14 - in the northern taiga, 15-21 - in the middle taiga).



Water acidification due to anthropogenic sulfate is characterized of ER.

In the acidic lakes of WS the water contained: chlorides, nitrates and sulfates.

Chlorides (lakes 13. 14. 20. 21) are dominated in majority lakes, but in some lake sulfates are dominated (lake 19).

Concentration of nitrates in water WS are higher in compared to the waters of ER.

Water of Western Siberia

NO_3^-

i) Delivery with the marsh waters, wetland and marsh is widely developed in the WS

($Nop_2 = 49.7 \cdot DOC - 114$. ($r=0.87$. $n=120$);

i) the gas flaring forms the nitrogen oxides;

Cl^-

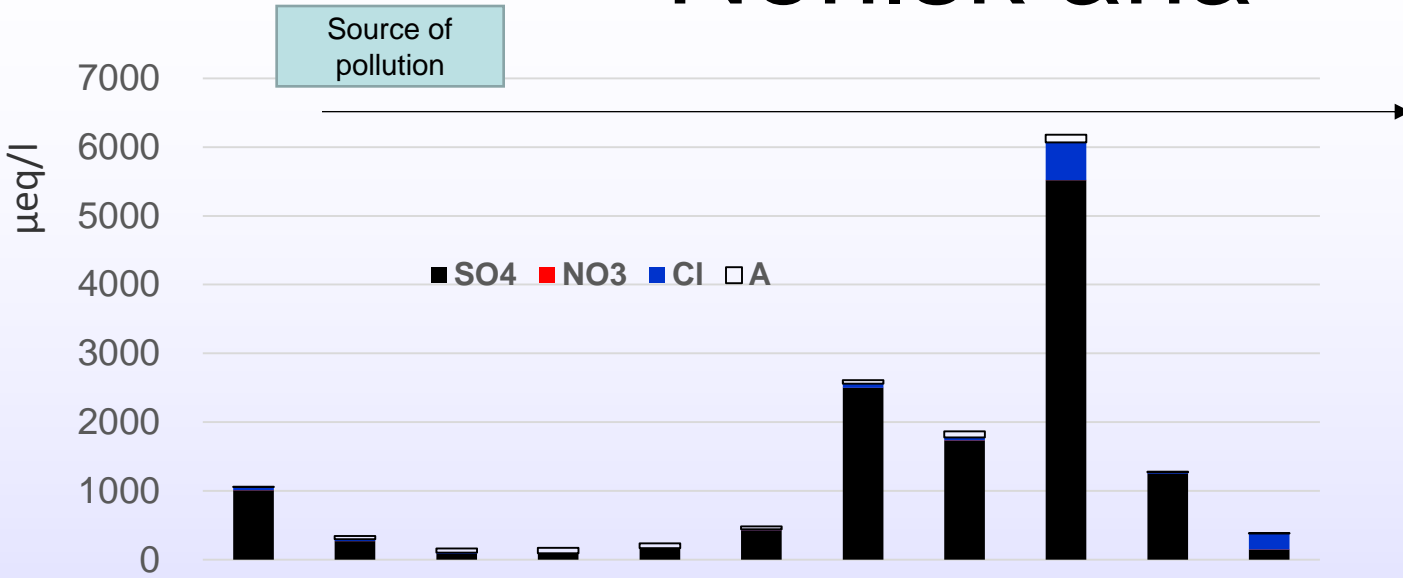
i) WS is located on the site of paleosee area and Quaternary rocks contain a amount of chloride (Arkhipov et al.. 1987).

ii) The chlorides are present in the waters of WS as part of pollution of the oil and gas fields development (Kiriushin et al.. 2013).

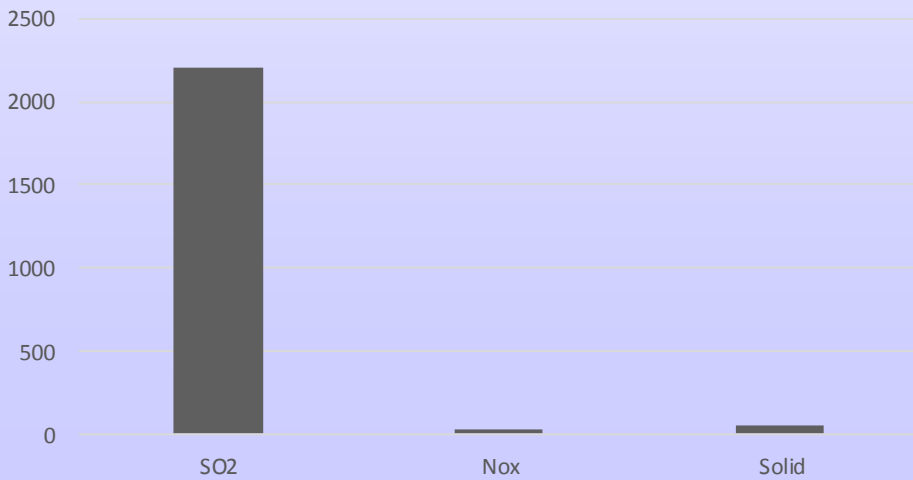
Organic compound (A^-)

Natural humus acids enters with the marsh waters.

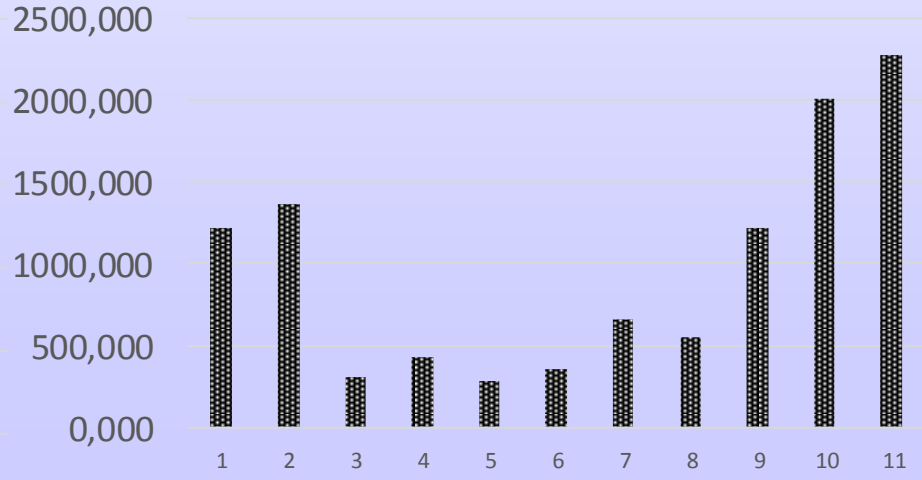
Norilsk aria



*1000 t/year



ANC, μeq/l



Forms specification



РОССИЙСКАЯ ФЕДЕРАЦИЯ



СВИДЕТЕЛЬСТВО

о государственной регистрации программы для ЭВМ

№ 2015617036

ACIDFORMMET

Правообладатель: *Федеральное государственное бюджетное учреждение науки Ордена Ленина и Ордена Октябрьской революции Институт геохимии и аналитической химии им. В.И. Вернадского Российской академии наук (ГЕОХИ РАН) (RU)*

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Заявка № 2015612317

Дата поступления 26 марта 2015 г.

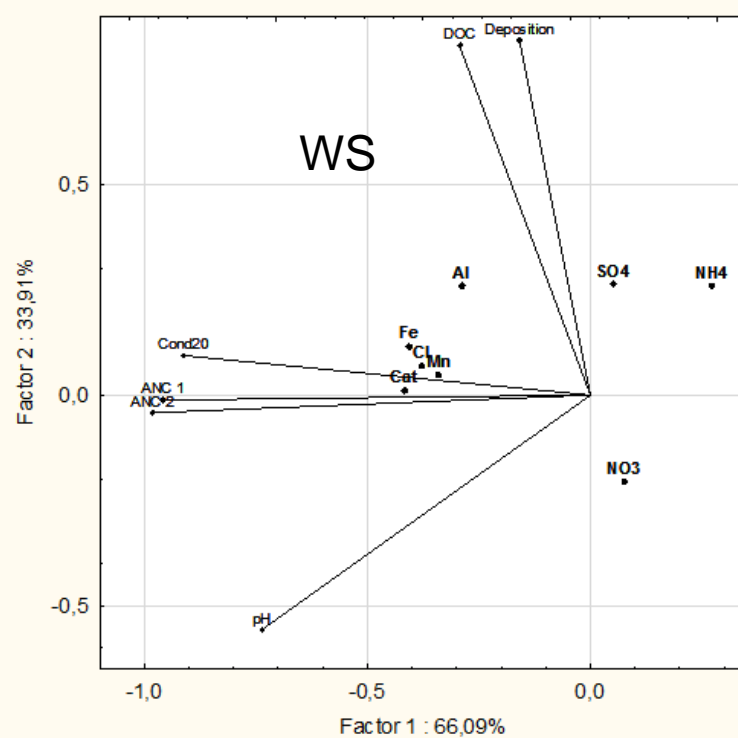
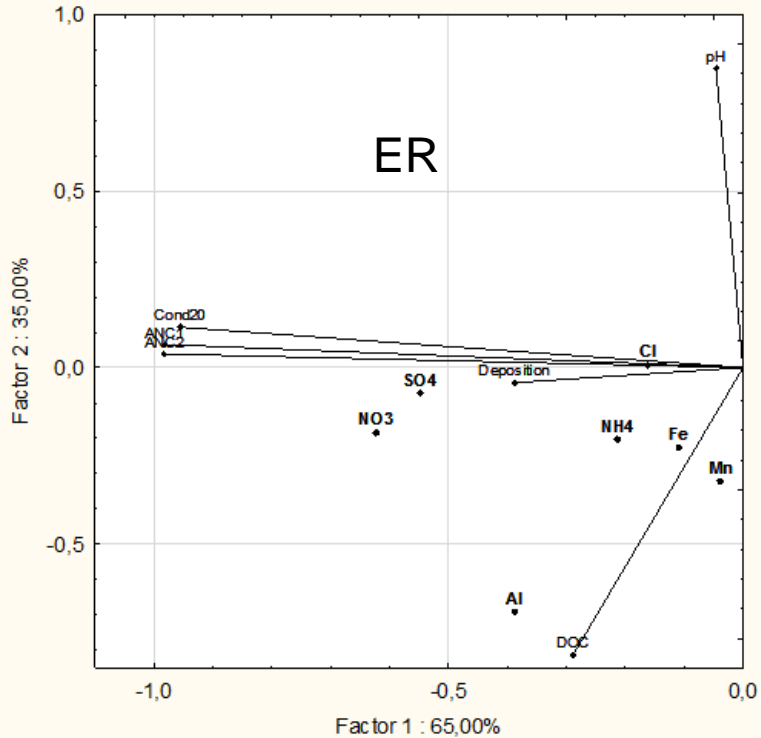
Дата государственной регистрации
в Реестре программ для ЭВМ 29 июня 2015 г.

Врио руководителя Федеральной службы
по интеллектуальной собственности

Л.Л. Кирий



In addition to conducting model experiments were performed calculations of the forms of metals (a technique Dinu Moiseenko. 2015) on the basis of data on the chemical composition of the water the lakes. Results showed high similarity - more than 80% for ions of iron, aluminum and heavy metals, the subgroup metals of chromium and tin, etc.

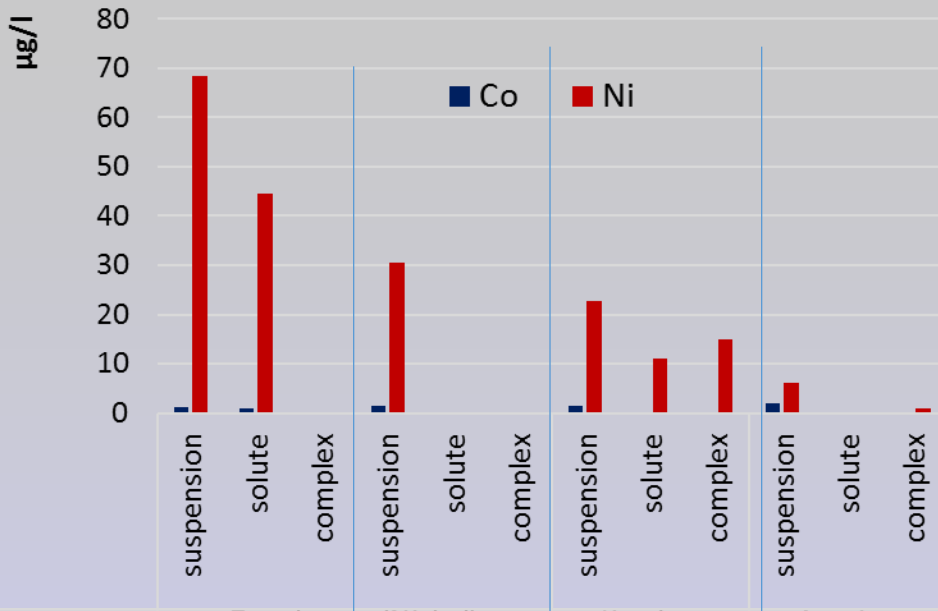


We have chosen same main explanatory variables and tested with used a PCA and RDA technique. According to the results of the calculation: **ANC 1 and ANC 2 are very close** that means the reliable of analyzes.

In ER: SO_4^{2-} , Cl^- and NO_3^- mostly have an affinity for Cond. In WS: SO_4^{2-} have an affinity for Deposition.

In WS water Cl^- have an affinity for Cond. But nitrate is inert to the selected axes. Despite the more higher sulfate concentrations in waters ER the technogenic sulfates also have a significant effect in the water lakes WS.

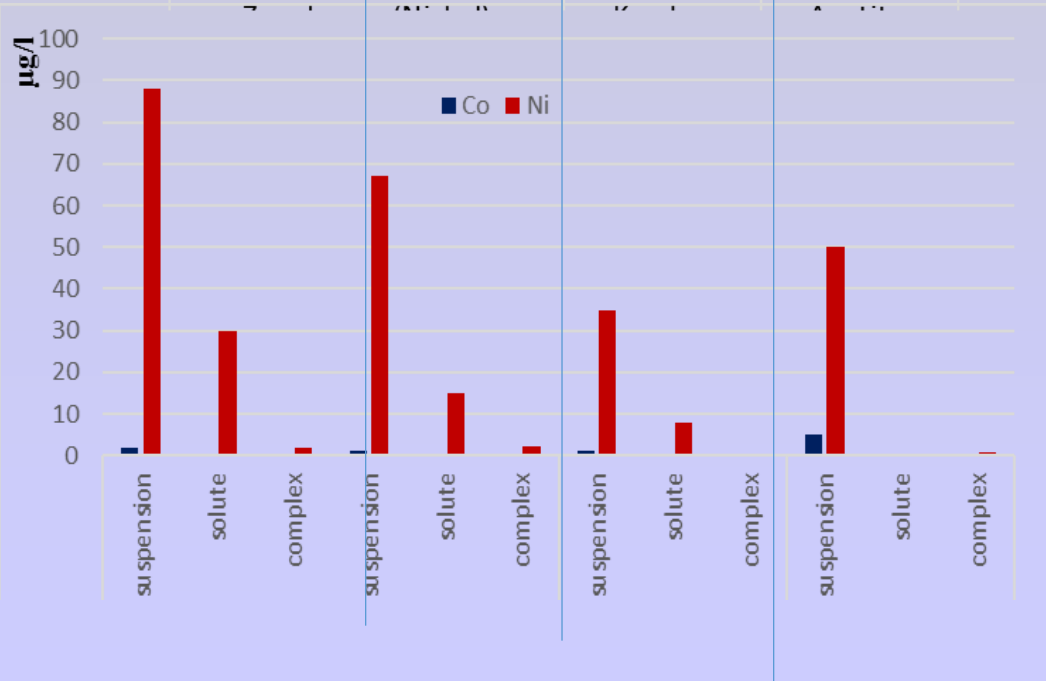
The greatest acid deposition in an areas with steady geological formations to acidification was found. This is reflected in synchronization of deposition increase and the water buffering to acidification towards the south.



Kola Lakes near source of pollution

The affinity of these elements to an organic substance as follows:

$\text{Fe} > \text{Al} > \text{Zn} > \text{Ni} > \text{Cu} > \text{Pb} > \text{La} > \text{Ce} > \text{Co}$



Norilsk Lakes near source of pollution

TOC (Mn) \ll 5 mg O/l

Degradation organic matter

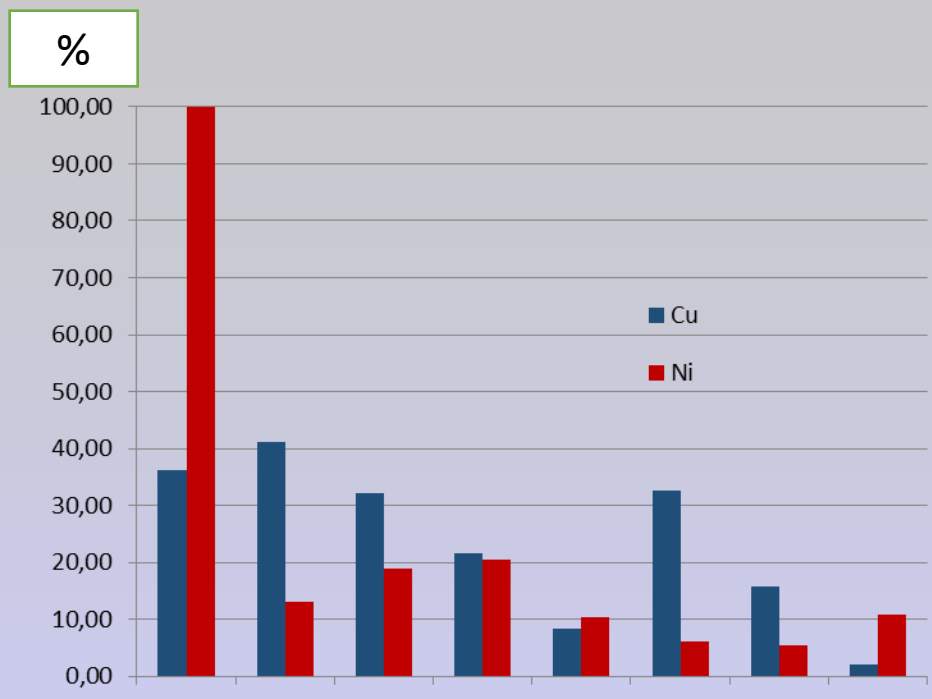
ANC \gg 1000 $\mu\text{eq/l}$

Very significant reduction of complex forms of nickel and copper, as well as their total contents was found.

The distribution of metal affinity for organic substances should be
 Lake1 - Lake 4
 Ni>Cu>Fe>=Al>Y

Lake 4 – Lake 8

Fe>Al>Cu>Ni>Y



Parameters	1	2	3	4	5	6	7	8
pH	6.50	6.55	6.25	6.18	6.64	5.99	6.16	6.48
Cond20	56	24	22	29	27	38	27	30
Color	58	18	30	51	28	55	67	62
COD Mn	10.54	3.76	6.87	9.29	5.58	13.74	11.00	10.67
Alk	316	85	73	140	151	67	116	156

Critical Loads and its exiding

$$CL = ([BC_o^*] - [ANC_{limit}]) Q - BC_d^*$$

$$BC_o^* = [BC^*]_t - F \Delta([SO_4^*] + [NO_3]) = [BC^*]_t - F (([SO_4^*]_t + [NO_3]_t) - ([SO_4^*]_o + [NO_3]_o))$$

$$F = \sin(\pi/2) [BC]_t / S$$

$$[SO_4]_{o^*} = 15 + 0.16[BC_t] \text{ * (Henriksen et al., 1992)}$$

European Russia

$$BC_o = [BC]_t - F (([SO_4]_t - [SO_4]_o))$$

Zones of tundra, forest-tundra and northern taiga (Kola Peninsula):

$$[SO_4^*]_o = 15.3 + 0.02 [BC^*]_t. r = 0.71. p < 0.001;$$

Zone of the middle taiga (Karelia):

$$[SO_4^*]_o = 15.4 + 0.11 [BC^*]_t. r = 0.64. p < 0.001;$$

Zone of mixed forests:

$$[SO_4^*]_o = 15.2 + 0.05 [BC^*]_t. r = 0.68. p < 0.001.$$

S is 400 $\mu\text{eq} / \text{l}$ for the tundra and taiga zones.

S is 1100 $\mu\text{eq} / \text{l}$ for mixed ER forests.

$$CL_{ex} = CL - SO_{4\text{dep}}^* - NO_{3\text{dep}} + BC_{\text{dep}}^*$$

Western Siberia

$$BC_o = [BC]_t - F (([SO_4]_t - [SO_4]_o) + ([NO_3]_t - [NO_3]_o) + ([Cl]_t - [Cl_{Na}]))$$

$$[NO_3]_o = 0.118 [A^{n-}]_t.$$

(Cl_{Na} is compensated by Na)

Zones of tundra, forest-tundra and northern taiga:

$$[SO_4]_o = 2.67 + 0.021 [BC]_t. r = 0.72. p < 0.001;$$

Zone of the middle taiga:

$$[SO_4]_o = 16.9 + 0.015 [BC]_t. r = 0.76. p < 0.001;$$

Zone of southern taiga:

$$[SO_4]_o = 12.4 + 0.002 [BC]_t. r = 0.69. p < 0.005.$$

S is 500 $\mu\text{eq} / \text{l}$ for the tundra and northern taiga zones.

S is 1250 $\mu\text{eq} / \text{l}$ for the middle taiga

S is 3000 $\mu\text{eq} / \text{l}$ for the southern taiga.

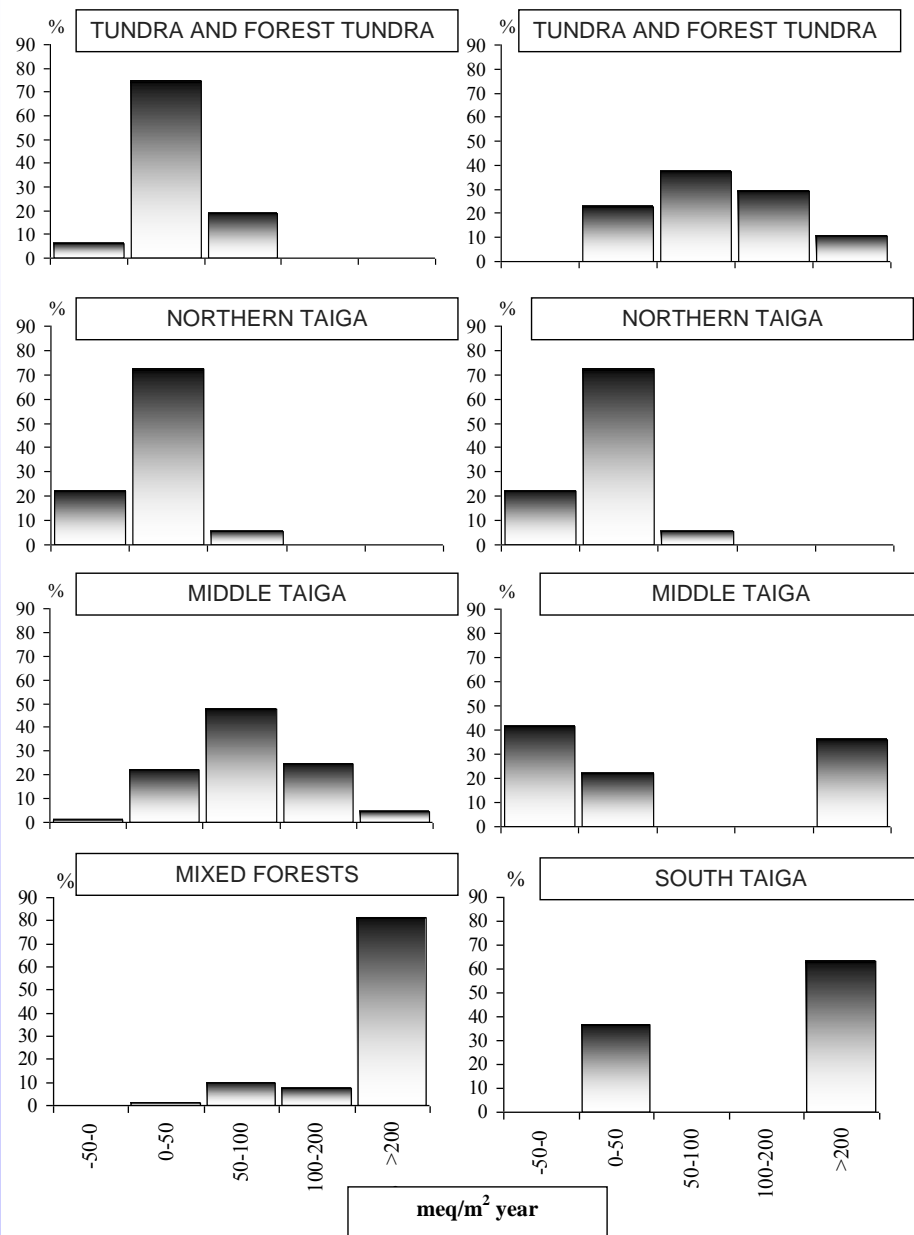
$$CL_{ex} = CL - SO_{4\text{dep}} - NO_{3\text{dep}} - Cl_{\text{dep}} + BC_{\text{dep}}$$

Thus, the necessary data for estimating the flow of cations into water systems ensuring neutralization of technogenic acids have been determined. Taking into account the complete and correctly obtained hydro-chemical information. ANClimit was adopted as 50, $\mu\text{eq} / \text{l}$

CL

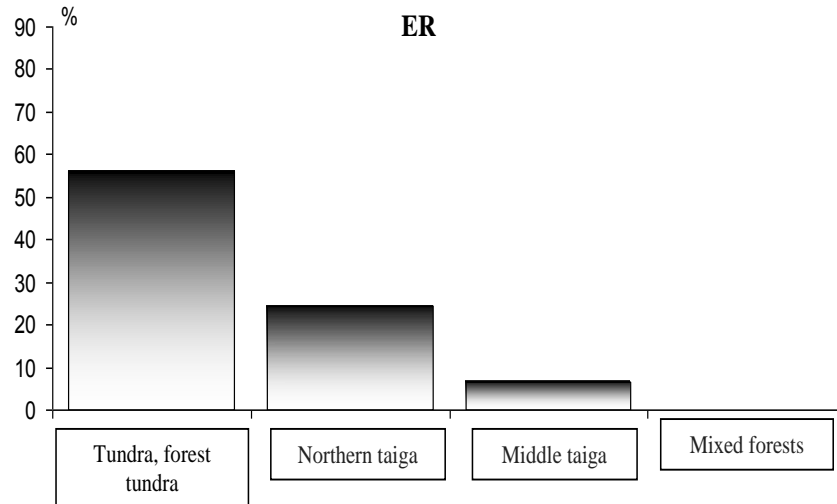
ER

WS

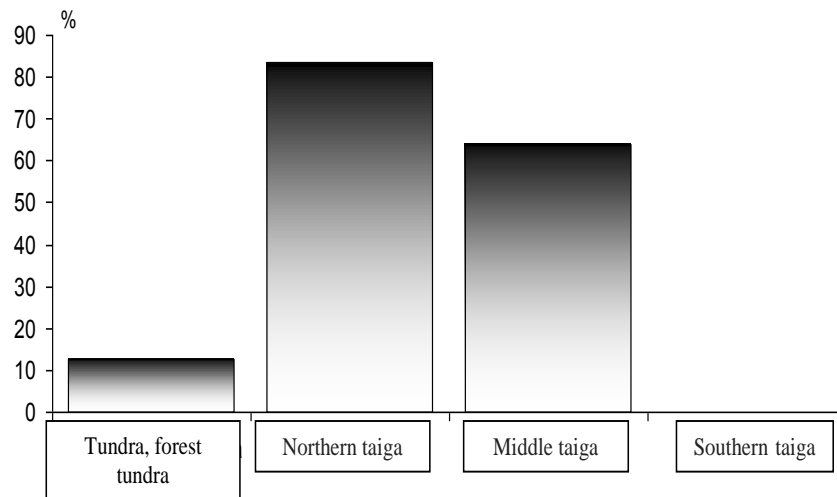


CLex

ER

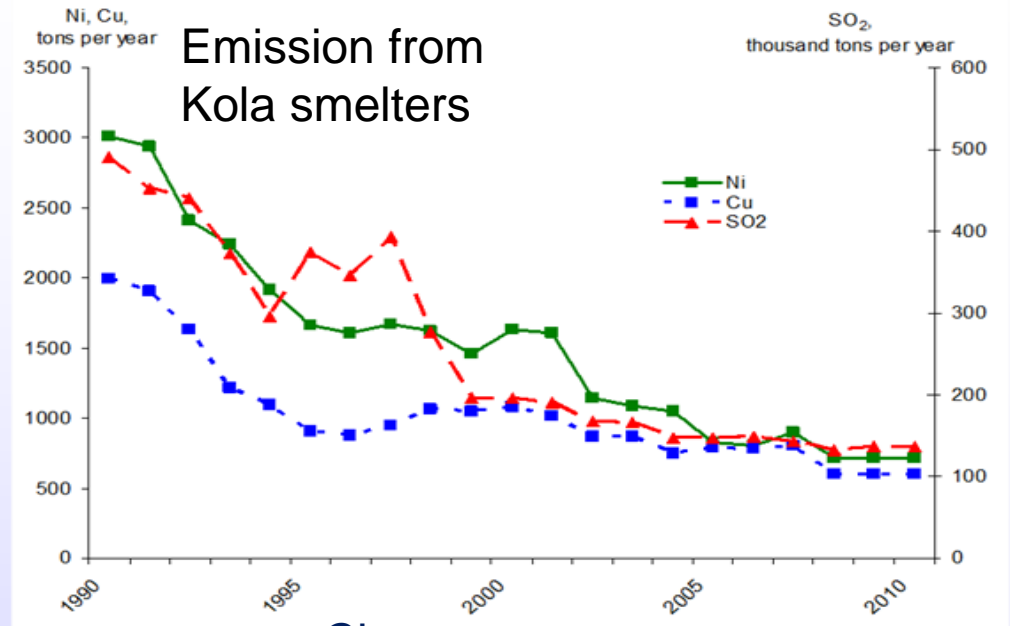
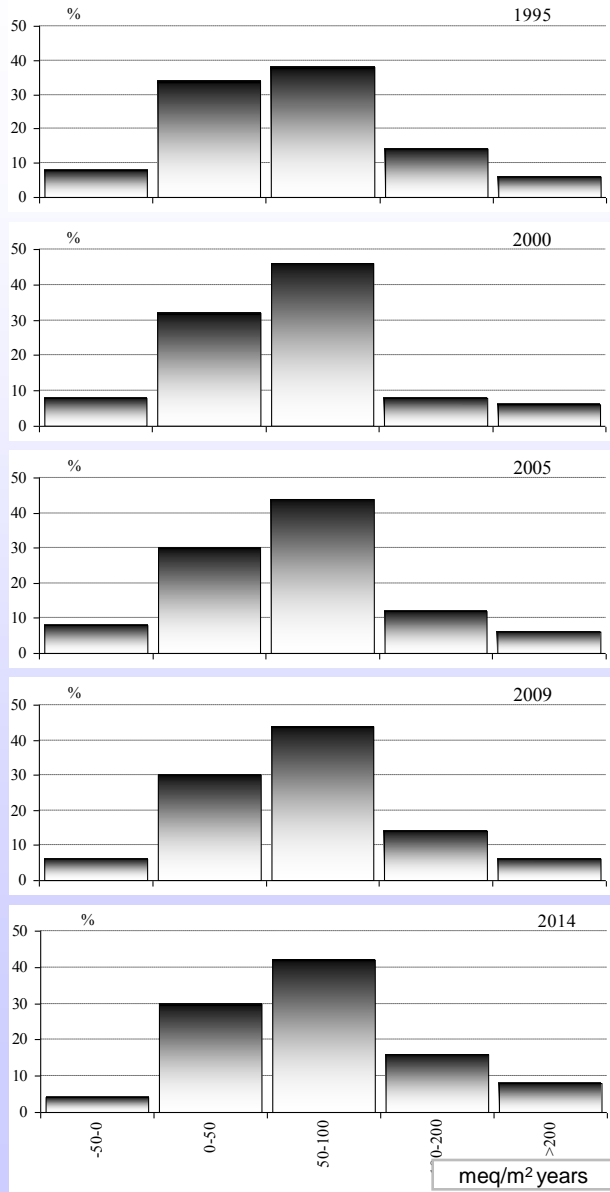


WS

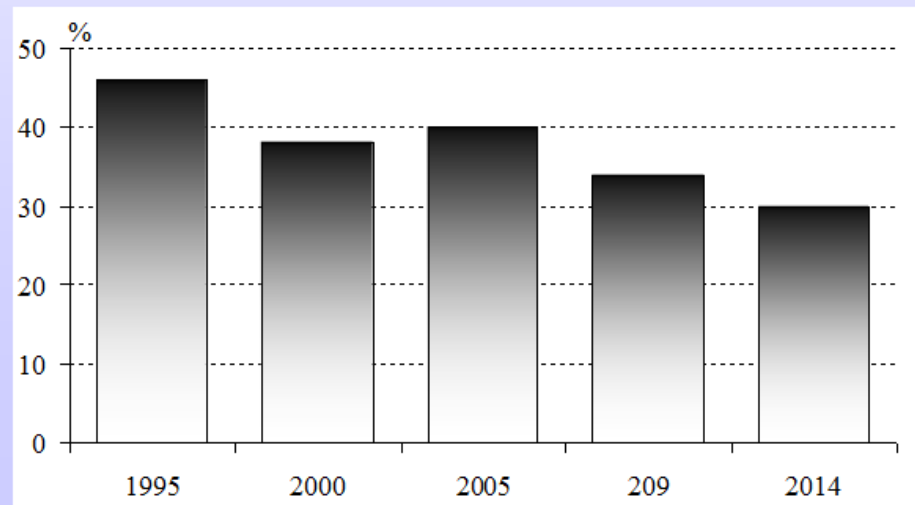


CL

Long-term trends- recovery



Clex



Conclusions

1. On the European territory of Russia, lakes with high buffer capacity are located in the taiga climatic zones, in Siberia - in the tundra regions. This is due to geochemical factors (geological rocks).
2. High anthropogenic influence (in the European territory of Russia - copper-nickel production in Western Siberia - oil refining complexes, in Eastern Siberia - Norilsk Nickel) determines the various chemical equilibria in natural waters.

For example, in the Norilsk region (from 0 to 500 km from the source), the absence of metal complexes with organic matter of water

3. The calculated critical loads and their exceedances for the European territory of Russia and Western Siberia showed the differences in the parameters for the climatic zones of the territories

**THANK YOU
FOR YOUR ATTENTION !**

